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# NEWGRAPHENEBATTERYCANBECHARGEDINJUST15MINUTES.

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#### Abstract:-

Anewtypeofbatteryhasbeendevelopedthatcouldrevolutioniseelectricvehicleusewitharechargetimeofjust15seconds. While technology is progressing in the electric bike sector on all these points, for most, the rate of progression is still just to slow. One factor that could improve the situation on two of the above mentioned fronts isrecharge time. Reducing this to shorter amounts of time improves the deficit that electrically powered motorcycleshave over their fossil fuels burning cousins. A battery with a recharge time of minutes, comparable to a petrol -powered bike, go some way to alleviating the range anxiety that comes with electric motorcycle ownership, and quickerrechargetimes are just what this new graphene battery achieves.

Scientistshavecreated agraphene 'super battery' that can be recharged in a round 15 seconds. The lithium —ion cellis also capable of surviving a claimed several hundred thousand charge cycles, increasing the longevity of the cellwhencomparedtoaconventionalbattery.

# Keywords:-

Battery, Capacity, Discharge, Efficiency, Electrodes, Graphene, Lithium-ion, Smartphones, Technology

#### Introduction:-

current

Graphene batteries are a new type of technology that allows for increased electrode density, faster cycle times, aswellaspossessingtheabilitytoholdthechargelongerthusimprovingthebattery's lifespan. Graphenebatteries are well established and come in many forms. Similar to graphite, there are now various types of functional graphenederivativeselectrodesandresearchersarediscoveringmultiplebenefits

whencompared to pure graphite electrodes. Our graphene battery user's guide includes four wellest ablished graphene electronest. odesdesignsofexperiments(DOE

) from cutting edge academic research including one for a Graphene-Lithium-Sulphur battery, the current frontrunner technology, and are included for reference. Graphene based batteries are being activity researched for manycommercial applications. The improved-performance and life-cycle advantages when developing graphenebasedbatteries over traditional metal-ion batteries are well worth the resource investment. Elon Musk's Tesla Motorsprovidesafamousexampleofinnovativecompaniesactivelypersuinggraphenebatteryresearchandcommercializa tion. We believe that the real graphene battery break through will be from graphene-lithium-ionhybrid chemistries cathodes of lithium-sulphur cells. incorporated into the This types of technology yearsawayfromcommercializationandintensiveresearchisongoing. The more innovative graphene battery technologies w ill require significant & D expenditures and will take many years to commercialized. Nowadays, zinc andalkaline based batteries are available, but they generally have a shorter lifespan due to their high charge density. Unlike lithium-based batteries, they can not operate at higher voltages. A primary (non-rechargeable) batteryconsists of two electrodes which allows the current to flow in one direction only, via an intermediary electrolyte. Secondary (rechargeable) batteries still consists of two electrodes however the lithium ions can flow in bothdirectionsdependingonifchargingordischarging. Theanodeisgenerallyalithiumbased (metaloxide)compoundandthecathodeaporouscarbon. Both the anodeand cathode have a rigid structure with defined holes, whichallowsfortheabsorptionoflithiumionsinto theholeswhenthecurrentisapplied. When thereisnocurrentbeingapplied, the ions desorb into the electrolyte solution. Absorption of the lithium ions can occur

ionabsorbintotheanode. This process allows for many cycles to be produced, learning to an enhanced life span.

on both the anode and thecathode, when a battery is in use, the ions move to the cathode. When charging, the reversed

the

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## Methodsandmaterials:-

## 1. Verticalalignedcarbonnanotubeelectrode.

NAWAtechnologyhasdesignedandpatentedanUltraFastCarbonElectrode,whichissaysitisagamechangerinthe battery market. It uses a Vertically-Aligned Carbon Nanotube (VACNT) design and NAWA says it can boostbatterypowertenfolds,increaseenergystoragebyafactorofthreeandincreasethelifecycleofabatteryfivetimes. The company sees electric vehicles as being the primary beneficiary, reducing the carbon footprint and cost ofbattery production, while boosting performance. NAWA says that 1000 Km range could become the norm, withchargingtimescutto5minutestogetto80percent.

# 2. Acobaltfreelithium-ionbattery

Researchers developed a lithium-ion battery that does not use cobalt for its cathode. Instead it switched to a highpercentage of nickel (89 percent) using manganese and aluminium for the other ingredients. "cobalt is the leastabundantandmostexpensivecomponentinbatterycathodes",saidoneresearcher.Researchersalsotoldthatwearecom pletely eliminating it. The team says they have overcome common problems with this solution, ensuring goodbatterylifeandanevendistributionofions.

# 3. Astepclosertosiliconanodelithium-ionbatteries.

Looking to overcome the problem of unstable silicon in lithium-ion batteries, researchers have developed a methodtoproduceahybridanode, using mesoporous siliconmic roparticles and carbon nanotubes. Ultimately the aimistore place graphite as the anode in batteries and use silicon, which has been ten times the capacity. Using this hybrid material improves the performance of the battery, while the silicon material is sustainably produced from barley huskash.

## 4. Lithium-sulphur-batteriescouldoutperformLi-ionhavelowerenvironmental.

Researchers have developed a lithium-sulphur battery that can power a smart phone for 5 days, out performinglithium-

ion. Theresearcherhavefabricated this battery, have patents and the interest of manufacture. The group has funding for further research in 2020, saying that continued research into cars and grid use will continues. The newbattery technology is said to have a lower environmental impact than lithium-ion and lower manufacturing costs, while offering the potential to power avehicle for 1000 Km (620 miles), or as mart phone for 5 days.

## 5. IBM'sbatteryissourcedfromseawaterandout-performancelithium-ion.

IBM research is reporting that it has discovered a new battery chemistry that is free from heavy metals like nickeland cobalt could potentially out-perform lithium-ion. IBM research says that this chemistry has never been used incombinationinabatterybeforeandthatthematerialscanbeextractedfromsea water.

#### 6. Panasonicbatterymanagementsystem:-

Panasonic says that its new technology can be easily applied with a change to the battery management. System, which will make it easier to monitor and evaluate batteries with multiple stacked cells, the sort of thing you mightfind in an electric car. Panasonic that this system will help the drive towards sustainability by being able to bettermanagereuseandrecyclingoflithium-ionbatteries.

## Resultsanddiscussions:-

Smart phone battery technology is pretty good these days. But if three is one thing gadget lovers will never be ableto get enough of it's the promise of better battery life. Would not it be great, if our handsets lasted two or three fulldays of heavy use with just a single charge? What about a whole week? With graphene batteries this might not besuchapipedream. Graphene batteries are not powering smartphone and other gadgets justyet, but the technology is

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progressing. In the future, graphene could be the material that replaces the lithium-ion batteries that the technologyhasbecomesoreliantonfor decades.

Keybenefitsofgraphenebatterytechnologyinclude:-

Smaller, slimmer battery, we have already discussed how graphene is light weight. It is when you stack 3 millionlayers of graphene is that you get 1 mm thickness. I mean, that should be enough to tell you that graphene batteries are not going to take much space in your future smart phone. It will allow manufactures to place higher capacity batteries in your phones, tablets, laptops, and more.

## Highercapacity:-

Graphene has a higher energy density as compared to lithium-ion batteries. When the latter is known to store up to 180 Wh per kilogramme. Graphene's capable of storing up to 1000 Wh per kilogrammeSo, you can have a highercapacitygraphene battery pack of the same size as the lithium-ion battery. Other benefits include faster chargingtimes, thermal management and grater safety. Graphene batteries are great for powering projects on breadboards, PCBs, and just about any circuit. They are very portable and come in a lot of form factors, with different voltage, current and power capabilities. If your circuit is in need of portable power or temporary power, batteries can be exactly what you need. Graphene batteries are said to be the absolute alternative to our current-gen. lithium-ionbatteries. Graphene batteries are itself quite lightweight, advanced and powerful. Graphene has been found to be asuperior material as it not only has higher electrical and heat conductivity, but it is also quite light weight, flexible and durable. Thus, graphene batteries have been under development for many years now and are expected gomainstreaminthenext coupleoftheyears. Graphene battery is a new technology, but it is does not meanthey have not been tested. Manufactures have dedicated quite sometime to graphene battery research and why would not they, especially when it's superior to the lithium-ion batteries. Graphene batteries have a number of benefits but the oneshortcomingthatisholdingitsmassadoptioninourdevicesismassproductionandcostsinvolvedinthesame. Whyis difficult to mass-produce graphenebatteries? It is because of the lack of a feasible technique for the massproduction of high quality graphene. You certainly could produce graphene at home using graphite and sticky tape, but that does not work for the mass production of the batteries. The lack of the same also drives up production costas quality of materials will need to be taken into account, which could as high as tends to thousands of dollars. Graphene batteries have extraordinary potential and yield results better than the existing battery packssomethingthat should have become quite clear to you by now. Research in this field has been quite rampant in the past coupleof decades, but we will still need to be patient for its commercialization. Many companies are currently restinggraphene batteries or are trying to improve lithium batteries with graphene to enhance their performance, but they are not fully commercially available at the moment.

Conclusions:-In terms of efficiency, graphene based batteries are quickly becoming comparable to conventional solid-state batteries. Thanks to the continuous advancement in graphene-based batteries, it will not be long beforethey outperform their soild-state predecessors. The additional advantages associated with graphene being present

intheelectrodescanbeuseful, eveniftheefficiency is notashigh. Graphene batteries are an ideal option for batteries that possess a similar efficiency and because of this reason researchers are striving to further advance this class of batteries. Graphene batteries have started to gain interest in the commercial marketplace and it will not be long before these batteries become the normand phase-out solid-state batteries. With ever-

increasingenergydemandsallacrosstheworld,developingimprovedenergystoragedeviceswithreducednegativeenviron mentalimpactsrelatedto consumer based battery usages is a noble objective and one that cheap tubes definitely supports.

Cheap

Tubeshopesthatthisguidehashelpedyoutoknowthecurrentgraphenebatteryresearchtrendsandinspiredyoutoinitiategraphenebatterydevelopment.

#### **References:-**

[1].AN,K.H,etal.Electrochemicalpropertiesofhighpowersupercapacitorsusingsingle-walledcarbonnanotubeelectrode.AdvFunct.Mater. 11,387-392(2001).

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- [2]. Cheng Q., Tang J., Zhang H., Graphene and carbon nanotube composite electrodes for supercapacitors with ultra-highener gydensity, Phy, chem., 2011, 13, 17615-17624.
- [3].C.N.R.Rao, A.K.Sood, R.Voggu, and K.S.Subramanyam, "somenovelattributes of graphene". J. phys. Chem., left. 1,572-580 (2010).
- [4].D.Wang,R.Kou.D.Choi,Z.Yang,Z.Nie,J.Lietal., "Ternaryselfassemblyoforderedmetaloxidegraphenenanocompositeforelectrochemicalenergy storage," ACSNano, 4,1587-1595 (2010).
- [5].G.Wang, X.Shen, J.Yaoand J.Park "Graphenenanosheets, forenhanced lithium in batteries". Carbon 47, 2049-2053 (2010).
- [6].HaegyomK.,Dong-
- HwaS.,SungwookNoneK.,KisukK.,HighlyreversibleCo304/graphenehybridanodeforlithiumrechargeablebatteries,carbon,2011,49(1),326-332.
- [7]. Huang X., Xiaoying Q., Boey F. and Zhang H., Graphene based composites, Chem. Soc. Rev. 2012, 41,666-686.
- [8].J.M.TarasconandM.Armand. "Issuesandchallengesfacingrechargeablelithiumbatteries". Nature (London) 414,359-367. (2001).
- [9]. "PanasonicDevelopesNewHigherCapacity18650graphenebattery, Application of silicon based alloyina node". Green carcongress.com Retrieved 3 1 January 2011.
- [10].Peres, N.M.R., Riberio, R.M. (2009). "Focuson graphene". New Journal of Physics." 11(9):095002.
- [11]. P. Lian. X. Zhu, S. Liang, Z. Li, W. Yang, and H. Wang, "Large reversible capacity of high quality graphene sheets as an anode material for lithium-ion batteries". Electrochim. Acta, 55,3909-3914. (2010).
- [12]. Stoller, M. D., Park, S., Zhu, Y., An, J. & Ruoff, R. S. Graphene -based ultracapacitors. Nano Lett. 8.3498-3502
- (2008).[13].Winter, M.&Brodd, RJ. Whatarebatteries, fuelcells and supercapacitors? chem. Rev. 104.4245-4270 (2004).